## A Five-Center Redox System: Molecular Coupling of Two Non-Innocent Imino*o*-benzoquinonato-Ruthenium Functions through a π Acceptor Bridge

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Combining the concepts of non-innocent behavior of metal/ligand entities and the coupling of redox-active moieties via an electronically mediating bridge led to the synthesis, the structural, electrochemical and spectroscopic characterization of  $[Cl(Q)Ru(\mu-tppz)Ru(Q)Cl]^{n}$ where  $O^{\circ}$ is 4,6-di-tert-butyl-N-phenyl-oiminobenzoquinone and tppz° is 2,3,5,6-tetrakis(2-pyridyl)pyrazine, the available oxidation states being  $Ru^{II,III,IV}$ ,  $Q^{o,-,2-}$  and  $tppz^{0,-,2-}$ . One-electron transfer steps between the n = (2) and (4+) states were studied by cyclic voltammetry and by EPR, UV-VIS-NIR spectroelectrochemistry for the structurally characterized anti isomer of  $[Cl(Q)Ru(\mu-tppz)Ru(Q)Cl](PF_6)_2$ ,  $2(PF_6)_2$ , the only configuration obtained. The combined investigations reveal that  $2^{2+}$  is best described as  $[Cl(Q^{-})Ru^{III}(\mu$  $tppz^{\circ}$ )Ru<sup>III</sup>(Q<sup>-</sup>)Cl]<sup>2+</sup> with antiferromagnetic coupling between the ruthenium(III) and the iminosemiquinone components at each end. Metal-based spin as evident from large g factor anisotropy (EPR) and an intense inter-valence absorption band at 1850 nm in the near infrared (NIR) suggest that oxidation occurs at both iminosemiquinones to yield two Ru<sup>II,III</sup>-bonded quinones, implying redox-induced electron transfer. Reduction takes place stepwise at the metal centers yielding iminosemiquinone complexes of Ru<sup>III,II</sup> as evident from radical complex EPR spectra with small <sup>99,101</sup>Ru hyperfine contributions. After complete metal reduction to ruthenium(II) the bridging ligand tppz is being reduced stepwise as apparent from typical NIR absorption bands around 1000 nm and from small g anisotropy of the monoanion  $[Cl(Q^{-})Ru^{II}(\mu-tppz^{-})Ru^{II}(Q^{-})Cl]^{-}$ .